Course Description Form

1. Course Name:

Organic Chemistry

2. Course Code:

ORCH105

3. Semester / Year:

FaII Semester / Academic Year 2023/2024

4. Description Preparation Date:

1-9-2023

5. Available Attendance Forms:

In Presence

6. Number of Credit Hours (Total) / Number of Units (Total)

2 hours Theoretical

3 hours practical /3.5 unit

7. Course administrator's name (mention all, if more than one name)

Name: Lecturer Sura Salim Hamid (Lecturer Alaa Taha Azeez

EamiI: surasaIimhamid74@uomosul.edu.iq

8. Course Objectives

Theoretical:

- Providing students with awareness of the importance of chemistry at the industrial, agricultural and environmental levels.
- Provide applications with a broad foundation and balance of knowledge and skills in organic chemistry.
- Developing the student's ability to apply their knowledge and professional skills in solving experimental problems in chemistry, which exceeds the goals of practical development.
- Developing the skills of valuable students in their field of specialization.
- Students gain from applying and employing their skills to serve society

Practical:

- Introducing and informing the student about the most important devices and equipment
- Used in the laboratory
- Introducing the student to the most important conditions that must be met in an ideal laboratory
- Introducing the student to safety procedures while working in the laboratory.
- Teaching the student the best diagnostic methods.
- Finding the appropriate and quick method for diagnosis
- Enable the student to perform calculations to find the concentrations of substances and the percentages of the resulting substances.
- Finding alternatives if the devices used are not available.

9. Teaching and Learning Strategies

Theoretical:

- Interactive lecture
- Brainstorming
- Dialogue and discussion
- Assignment of reports
- Conduct daily tests and monthly examinations

Practical:

- Interactive lecture
- Discussion, dialogue and brainstorming
- Conducting laboratory experiments
- Set reports
- · Conduct daily tests and
- Monthly checks

10. Course Structure

Week	Hours	Required Learning Outcomes	Unit or subject name	Learning method	Evaluation method
1	2h 3h	A1: The student is introduced to the concept of organic chemistry and its importance in different areas of life. B6: Student sets the melting point	Theoretical: General principles of organic chemistry practical: Determination of melting point	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
2	2h 3h	A2: The student is familiar with the most important properties, names, reactions, and preparation of alkanes B7: Write a report on the boiling point	Theoretical: Saturated Hydrocarbons (alkanes) practical: Determination of boiling point	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
3	2h 3h	B1: The student enumerates the types of alkenes in terms of nomenclature and metho of preparing them A7: The student uses a distillation device for purification	Theoretical: Unsaturated Hydrocarbons (alkenes) practical: Purification of liquid organic compounds by simple distillation	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
4	2h 3h	B2: The student explains the types of reactions of alkenes and dienes A8: The student learns about the types of solvents used for	Theoretical: Reactions of alkenes and types of dienes Practical: Recrystallization + Scientific visit	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions

		recrystallization			
5	2h 3h	A3: The student learns about the types of alkynes in terms of nomenclature, methods of preparing them, and their reactions C5: The student participates in the purification of solid organic compounds by sublimation	Theoretical: Alkynes (acetylenes) practical: Sublimation	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
6	2h 3h	A4: The student distinguishes between the chemical and physical properties of aromatic compounds and ways to name them B8: The student carries out a practical application procedure on how to separate liquid or solid organic compounds by solvent extraction	Theoretical: Properties and nomenclature of aromatic compounds practical: Solvent extraction	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
7	2h 3h	C1: The student distinguishes between the chemical and physical properties of aromatic compounds and ways to name them A9: The student memorizes the method for preparing methan gas in the laboratory	Theoretical: Preparation and reactions of aromatic compounds practical: Preparation of methane gas	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions

8	2h 3h	B3: The student explains the properties and names of alcohols and phenols A10: The student learns how to prepare 1_Butene	Theoretical: Properties and nomenclature of alcohols and phenols practical: Preparation 1_ Butene	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
9	2h 3h	B4: The student distinguishes between the methods of preparation and reactions of alcohols and phenols B9: The student applies the preparation of acetylene	Theoretical: Preparation and reactions of alcohols and phenols practical: Preparation of acetylene gas	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
10	2h 3h	A5: The student learns about ethers, how to prepare them, and the types of their reactions B10: The student masters the practical application of detecting types of alcohol	Theoretical: Ethers practical: Study of the properties of alcohols	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
11	2h 3h	B5: The student masters how to name, prepare and react aldehydes A11:The student differentiates between aldehydes and ketones through a practical application	Theoretical: Preparation, naming and reactions of aldehydes practical: Reaction and detection of aldehydes and ketones	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions
12	2h 3h	C2: The student emphasizes the names, preparation, and reaction of ketones C6: The student participates in a practical application on how to prepare acetone	Theoretical: Preparation, nomenclature and reactions of ketones practical: Preparation of acetone	Lectures And audio means And reports And conduct experiments	Exams Reports Discussion and questions

13	2h	A6: The student	Theoretical:	Lectures	Exams
	3h	participates in a	Properties and	And audio	Reports
		practical application	nomenclature of	means	Discussion and
		on how to prepare	carboxylic acids	And reports	questions
		acetone	practical:	And	
		B11: The student	Preparation of	conduct	
		experiences a practical	propanoic acid	experiments	
		application on how to			
		prepare			
		Propanoic acid			
14	2h	C3: The student	Theoretical:	Lectures	Exams
	3h	participates in the	Reactions and	And audio	Reports
		types of reactions and	preparation of	means	Discussion and
		methods of preparing	carboxylic acids	And reports And	questions
		carboxylic acids C7: The student	practical: Preparation of	conduct	
		understands how to	propionaldehyde	experiments	
		prepare	propronatuenyue	experiments	
		propanealdehyde			
15	2h	C4: The student	Theoretical;	Lectures	Exams
15	3h	understands the	Amines	And audio	Reports
	311	importance of amines	practical:	means	Discussion and
		C8: The student	Detect items	And reports	questions
		participates in methods		And	•
		for detecting elements		conduct	
				experiments	
		Evaluation			
			Evaluation date (one	Grade	Relative
1	T. 1		week)	7.1	weight %
	Final theoretical report +		Theoretical 15 weeks	7theoretical +	13%
	Short test	•	Practical 1-15 weeks 3 weeks	6 practical +	6%

1.	l.Course Evaluation		T	
t	Evaluation methods	Evaluation date (one	Grade	Relative
		week)		weight %
1	Final theoretical report +	Theoretical 15 weeks	7theoretical +	13%
	theoretical practical reports	Practical 1-15 weeks	6 practical	
2	Short test 1 Quiz	3 weeks	4theoretical +	6%
			2practical	
3	Midterm exam (theoretical and	9 weeks	10theoretical +	15%
	practical)		5 practical	
4	Short test 2 Quiz	12 weeks	4 theoretical +	6%
			2 practical	
5	Final practical test	practical exams week	20	20%
6	Final theoretical exam	theoretical exams week	40	40%
			100	100

12.Learning and Teaching Resources

Required textbooks (curricular books, if any)	Organic Chemistry book Authors:
	 Prof. Dr. SaIim Hamid Hussein
	 Prof. Dr. Sami Abdul-Ali
	KhaIId FathI AI ShaharI

	University of Mosul 2013 Dar Al-Kutub for Printing and Publishing
Main references (sources)	Organic Chemistry Authors:
Recommended books and references (scientific journals, reports)	Principles of Organic Chemistry
	Authors: • Prof.Dr. Mohamed Magdy Wasel/Cairo
	Fundamentals of Organic Chemistry
	Authors:
	Prof. Dr. Mohamed Wasel
Electronic References, Websites	https://arabian-chemistry.com/ https://scholar.google.com/

Lecturer Name (Theory)	Lecturer Name (Practical)
Lecturer Sura Salim Hamid	Lecturer AIaa Taha Azeez
Head of Department of Forestry	Chairman of the scientific committee
Dr Muzahim Saeed Al-Bek	Prof. Dr. Muhammad Younis Al-Allaf