Course Description Form

1. Course Name:
ANALYTICAL Chemistry

3. Course Code:

ANCH107

5. Semester / Year:

Spring semester 2023

7. Description Preparation Date:

9.

1-2-2024

11. Available Attendance Forms:

attendance

13. Number of Credit Hours (Total) / Number of Units (Total)

75h /3.5 unit

15. Course administrator's name (mention all, if more than one name)

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Course Objectives

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Enabling students to know the principles of devices

Identify the characteristics of the devices

Accurately

Finding the best methods for analysis

Finding the appropriate and quick

Method for analysis

Enable the student to perform

calculation

To find concentrate the analyzed

Materials and compare them with

Standard methods

Finding alternatives if the devices used

Are not available

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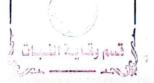
19. Teaching and Learning Strategies

Strategy

Applying modern strategies for Education
Providing learners with many different skills and knowledge increase students ability to learn using effective modern strategies that help
.Assigning group work to reveal
.Leadership skills
Assigning tasks and reporting
For each experiment

21. Course Structure

Week	Hours	Required Learning Outcomes	Unit or subject name	Learning method	Evaluation method
1	2h 3h	a2The student gets to know What is meant by b Chemistry Analytical /practical a2The student blames him On the app Measures related to the concept Ways and means To use devices	Introduction to chemistry Analytical Practical /guidelines About working in the laboratory	Lectures And Means Audio And Reports And other method	Disucussions
2	2h 3h	b6The student masters methods Expression Abou t focus and preparation Solutions Practical /b6 masters the laws used To prepare solutions	Ways of expression About focus and preparation Solutions Practical/laws used To prepare solutions Mathematical examples	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions
3	2h 3h	a3 Proficient in solvingmathematical examples Practical preparation Solutions Practical	And an introduction to Analytical chemistry Practical preparation Solutions Practical mathematical examples	Lectures And Means Audio And Reports	Exams Reports Disucussions Kuzat



		b3:The student carries out a practical application To prepare standard	experiment standard	And Reports And other	
_	1	acid a2:LThe student gets to	Oxidation erosion	method	Exams
6	2h 3h	know Redox modifications	And shorthand Practical/preparation	And Means Audio	Reports Disucussions
		Practical b3: 1The practical application carries	experience Standard base	And Reports	
		out a preparation experiment Standard base		And other method	T
7	2h	a2:The student gets to know	Complex formation studies Practical	Lectures And	Exams Reports
	3h	Analysis of complex formation Practical	/ iron estimation experiment with Potassium permanganate	Means Audio And Reports	Disucussions
		b3: A practical application carries out an	U. 1 E HT	And other method	
		estimation experiment Iron(II) with permangar		risk v ^{ill} tr	
В	2h	b6:The student gets to know	Depositional facies Practical / iron estimation	Lectures And	Exams Reports
	3h	Depositional facies Practical b3: Performs a	experiment With potassium dichromate	Means Audio And	Disucussions
		practical application Iron estimation experiment	: 07'	Reports And other method	
		With potassium dichromate		عة المدمل	J 200
	2h	a5:The student learns	Weight analysis	Lectures	Exams

	3h	about analysis Al-Wazani And the differences with Depositional delamination Practical a2: The student gets to know Testimonials Formation of complexes	And the differences with Depositional delamination PARTICAL / corrections Formation of complexes	And Means Audio And Reports And other method	Reports Disucussions kuzat
10	2h 3h	a5: The student learns about analysis The mechanism theories that She came for him Practical c: A practical Application implements a calcium Determination experiment In chalk Using corrections Formation of complexes	Instrumental analysis and theories that She came for him Practical/experiment for calcium determination In chalk Using corrections Formation of complexes	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions t
11	2h 3h	b4: The student learns about measurement methods chromatographic analysis Practical c: A practical Application implements an estimation experiment Total hardness o f water Using EDTA	Measurement methods in Color analysis Practical/ experience hardship assessment College water using EDTA	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions
12	2h 3h	a2He knows with appreciation Selected chemicals Practical b3: A practical application implements estimation experiment Chloride by Moore's method in salt the food	To estimate Selected chemicals Practical/experiment for chloride estimation Murphy's table salt method	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions kuzat
13	2h	a6:The student gets to	Atomic absorption	Lectures	Exams

	3h	know Atomic Absorption spectrometry Practical b3:A practical application implements estimation experiment Chloride by Moore's method in drinking water	spectrometry Practical/ assessment experience Chloride by Moore's method Drinking water	And Means Audio And Reports And other method	Reports Disucussions
14	2h 3h	a2:The student is familiar with preparation methods Samples For chemical analysis Practical b3:A practical application implements an estimation experiment Chloride by Volhard's method salt	Sample preparation methods For chemical analysis Practical /assessment experience Chloride by the Volhard method In table salt	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions
15	2h 3h	b6:The student is proficient in solving open-ended questions Analytical chemistry Practical b6: The student masters various questions about Practical chemistry and its experiments	Open questions in Analytical chemistry practical/ Various questions about Practical chemistry and its experiments	Lectures And Means Audio And Reports And other method	Exams Reports Disucussions

23. Course Evaluation

Calendar methods	Calendar appointment(week	class	Relative weight%
Final report(experiments+practical	Theoreticalweek15 Practical week1-15	practical6+theoretical7	%13
Short test1	Week3	practical2+4theoretical	%6
Midtermtheoretical+practicalexam	Week9	practical5+10theoretical	%15
Short test2	Week12	practical2+4theoretical	%6
finalpracticaltest	Practical exam week	20	%20
Final theoretical test	Theory exam week	40	%40
The total		100	%100

Required textbooks (curricular books, if any)	Quantitative of inorganic chemistry by Vogel,1973.	
Main references (sources)	General chemistry for students of the College of Agriculture and Forestry Written by Dr. Sami Abdel Ali, Dr. Salem Hamed, Dr. Moaz Abdullah	
Recommended books and references	Foundations of analytical chemistry Dr., Thabet Al-Ghabsha, Dr. Muayad Qasim Al-Abayi	
(scientific journals, reports)	Dia Madeerii Giadona, 21 Madyaa Qasiii Ili Medy	

Theoretical subject teacher Farah Sameer salh	practical subject teacher MRS.ALAA TAHA AZEEZ
Head of the department	Chairman of the cientific committee

