

## Course Description Form

1. Course Name:	
Biochemistry	
2. Course Code:	
AGPP24_F3131	
3. Semester / Year:	
First semester (fall) 2024-2025	
4. Description Preparation Date:	
1/2/2024	
5. Available Attendance Forms:	
Presence	
6. Number of Credit Hours (Total) / Number of Units (Total)	
75 hours / 3.5 units	
7. Course administrator's name (mention all, if more than one name)	
Name: Dr.Arkam Mohamad Alomary and Afkar Yahya Ahmed	
8. Course Objectives	
<p>Theoretical</p> <ul style="list-style-type: none"> <li>- To enable students to know the importance of the basic principles of Biochemistry</li> <li>- The importance of the biochemistry program in everyday life and the economic and medical importance of this program</li> <li>_ Finding the best means to explain the proposed program and identify the characteristics of the devices accurately</li> <li>_ How to use modern technology and technologic machines to improve and develop the proposed program</li> <li>_ Enable the student to apply and employ this program and use it as one of the most important criteria for future employment in the communit</li> <li>_ Finding civil and governmental organizations or specializationsWho is interested in this program and how to connect or recruit students</li> <li>By understanding the concepts of Life Technologie</li> </ul>	<p>Practical</p> <ul style="list-style-type: none"> <li>- Enable the student to get acquainted with the principles and methods recent developments in the study of life chemistry Sciences</li> <li>- Study of carbohydrate synthesis</li> <li>Fats, proteins and all the tests that</li> <li>Conducted and disclosed</li> </ul>
9. Teaching and Learning Strategies	
<p>Theoretical</p> <ul style="list-style-type: none"> <li>- Interactive lecture .</li> <li>-dialogue and discussion .</li> </ul>	<p>Practical</p> <ul style="list-style-type: none"> <li>Interactive lecture .</li> <li>-dialogue and discussion .</li> </ul>

-Assignment of tasks and report			-Assignment of tasks and report - Practical tests conducted in the laboratory.		
10. Course Structure					
Week	Hours	Required Learning Outcomes	Unit or subject name	Learning method	Evaluation method
1	2Theoretical 3Practical	<b>THEORETICAL</b> <b>b1</b> Shows the concept of water And its relation to properties Chemist and Physicist  <b>PRACTICAL</b> <b>b1</b> Applies rules Safety specifications in Laboratories	<b>THEORETICAL</b> Water  <b>PRACTICAL</b> Safety in laboratories	<b>THEORETICAL</b> Lectures Audio media Reports and other methods  <b>PRACTICAL</b> Follow Instructions Found in laborator	discussions
2	2Theoretical 3Practical	<b>THEORETICAL</b> <b>c1</b> Shows the most important differences in composition Carbohydrate chemis  <b>PRACTICAL</b> <b>a1</b> Classify carbohydrates Types and type	<b>THEORETICAL</b> Carbohydrates  <b>PRACTICAL</b> Carbohydrates and th types	<b>THEORETICAL</b> Lectures Audio media Reports and metho Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
3	2Theoretical 3Practical	<b>THEORETICAL</b> <b>b2</b> Familiar with the influencing factors For amino acids  <b>PRACTICAL</b> <b>b2</b> Solubility test is applied and the mulch test	<b>THEORETICAL</b> Amino acids: And peptides  <b>PRACTICAL</b> The tests General principles of carbohydrates	<b>THEORETICAL</b> Lectures Audio media Reports and metho Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
4	2Theoretical 3Practical	<b>THEORETICAL</b> <b>a1</b> Recognizes the mechanics of the acti of proteins inside the body	<b>THEORETICAL</b> Protein  <b>PRACTICAL</b>	<b>THEORETICAL</b> Lectures Audio media Reports and metho Other	Exams reports discussions

		<b>PRACTICAL</b> <b>a2</b> Recognizes carbohydrate reductive tests	The test Carbohydrates	<b>PRACTICAL</b> Procedure Tests Operation	
5	2Theoretical 3Practical	<b>THEORETICAL</b> <b>c2</b> It shows the changes that you get into Felts  <b>PRACTICAL</b> <b>b3</b> Tests types of carbohydrate description	<b>Theoretical</b> lipidis  <b>PRACTICAL</b> Hydrolysis of sucrose and iodine testing and hydrolysis of starch with mineral acids	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
6	2Theoretical 3Practical	<b>THEORETICAL</b> <b>c3</b> Suggests a suitable method For the action of enzymes  <b>PRACTICAL</b> <b>b4</b> Performs lipid-specific tests	<b>THEORETICAL</b> Enzymes  <b>PRACTICAL</b> Special tests With fat	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	Exams reports Quiz discussions
7	2Theoretical 3Practical	<b>THEORETICAL</b> <b>c4</b> Notice the changes That you get into the main article and his work with enzymes  <b>PRACTICAL</b> <b>a3</b> Selects fat constants	<b>THEORETICAL</b> Basic material  <b>PRACTICAL</b> Basic material Acrolein testing For the detection of glycerol	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
8	2Theoretical 3Practical	<b>THEORETICAL</b> <b>a1</b> Recognize the most important Changes that enzymes  <b>PRACTICAL</b> <b>a2</b> Understands the pH	<b>THEORETICAL</b> Mechanical work Enzymes  <b>PRACTICAL</b> PH is the pH value	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	Exams reports Quiz discussions

9	2Theoretical 3Practical	<b>THEORETICAL</b> <b>b2</b> Judging the efficiency of the work Nucleic acids Inside the body  <b>PRACTICAL</b> <b>a2</b> Specifies general and Descriptive Tests for amino acids	<b>THEORETICAL</b> Nucleic acids   <b>PRACTICAL</b> General tests Description distribution of Acids Aminism	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	Exams reports Quiz discussions
10	2Theoretical 3Practical	<b>THEORETICAL</b> <b>a2</b> Identify the most important materials Chemical constituent of DNA  <b>PRACTICAL</b> <b>b2</b> Explains the methods of detection of sulfur-containing amino acids	<b>THEORETICAL</b> DNA   <b>PRACTICAL</b> Acid detection Amino acids containing Sulfur	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
11	2Theoretical 3Practical	<b>THEORETICAL</b> <b>b4</b> Mastered the ingredients The main types of RNA nucleic acids and their types  <b>PRACTICAL</b> <b>a6</b> He tries the Mello test and the xanthoproteic test	<b>THEORETICAL</b> RNA   <b>PRACTICAL</b> Tests Amino acids:	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
12	2Theoretical 3Practical	<b>THEORETICAL</b> <b>e1</b> Defines the components of Basic nucleosides  <b>PRACTICAL</b> <b>c1</b> Recall Descriptive Tests for proteins	<b>THEORETICAL</b> Nucleosides   <b>PRACTICAL</b> The tests Description distribution of proteins	<b>THEORETICAL</b> Lectures Audio media Reports and methods Other <b>PRACTICAL</b> Procedure Tests Operation	Exams reports Quiz discussions
13	2Theoretical	<b>THEORETICAL</b>	<b>THEORETICAL</b>	<b>THEORETICAL</b>	Exams

	3Practical	<b>a2</b> Recognize the most important Biological qualities For nucleic acids  <b>PRACTICAL a2</b> The puritanical test is called	Biological qualities For nucleic acids  <b>PRACTICAL</b> The purite test	Lectures Audio media Reports and metho Other  <b>PRACTICAL</b> Procedure Tests Operation	reports Quiz discussions
14	2Theoretical 3Practical	<b>THEORETICAL b5</b> Familiar with the sources of composition and vitamin damage  <b>PRACTICAL a3</b> Characterizes the precipitation of prote with salts of heavy metals	<b>THEORETICAL</b> Vitamins  <b>PRACTICAL</b> Precipitation of prote Salts of heavy metals	<b>THEORETICAL</b> Lectures Audio media Reports and metho Other  <b>PRACTICAL</b> Procedure Tests Operation	Exams reports Quiz discussions
15	2Theoretical 3Practical	<b>b5</b> Solves a problem	A scientific visit to one Life chemistry laboratories	Getting to know One-to-One devices and tools f laboratories Life chemistry	Write a report on devices and tests that were identified during scientific visit

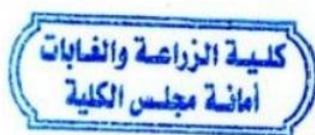
### 11. Course Evaluation

t	Evaluation methods	Evaluation date (one week)	Grade	Relative weight %
1	Final theoretical report + theoretical practical reports	Theoretical 15 weeks Practical 1-15 weeks	7theoretical + 6 practical	13%
2	Short test 1 Quiz	3 weeks	4theoretical + 2practical	6%
3	- Semester exam 1 (theoretical and practical) - Semester exam 2 (theoretical and practical)	5 weeks 10 weeks	10theoretical + 5 practical	15%
4	Short test 2 Quiz	12 weeks	4 theoretical + 2 practical	6%
5	Final practical test	practical exams week	20	20%
6	Final theoretical exam	theoretical exams week	40	40%
			100	100

### 12. Learning and Teaching Resources


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<b>Theoretica</b> Required textbooks (curricular books, if any) Theoretical approach to the course of principles Biotechnology / counter: D. Faten Dhawi Al-Muhanna/ PhD in philosophy Biochemistry science Department of Molecular Biology and Biotechnology	<b>Practical</b> The basics of Biochemistry Basil Kamel Al-Dlaly
Main references (sources) A library , scientific sites on the internet, access lectures for other Iraqi universities.	Basics of biochemistry by Basil Kamel Al-Dlaly
Electronic References, Websites	The World Health Organization(WHO) , the US Food and drug organization (USFDA)



  
الدكتور  
فاغن د. فغن  
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أ.د. هيلين السو محم  
رئيس اللجنة العلمية