urso Description Form $\mathbf{\Omega}$

	Course Description Form						
1. Co	ourse Name	2:					
Pharm	Pharmaceutical Calculations (Theoretical+ Practical)						
2. Co	ourse Code	:					
Phind	23 1212						
3. Se	emester / Y	ear:					
2 nd Se	mester/1 st	year					
4. De	escription I	Preparation Date:					
25/2/2	25/2/2024						
5. Available Attendance Forms:							
Stude	Students' signature on attendance sheet						
6. Ni	umber of C	redit Hours (Total) / Nu	mber of Units (Tota	l)			
2 hour	rs Theoreti	cal + 2 hours Practical (6	50) /3 units				
7. Co	ourse admin	nistrator's name					
		Т	Theoretical				
Name: As	sist. Lec. N	Mohmmed Khalid Al-Sh	aheen				
Email: m	ohammed.k	<u>chalid@uomosul.edu.iq</u>					
			Practical				
Name: As	sist. Lec. A	Alaa Rakan Al-Taie					
Email: <u>ala</u>	<u>altaie@</u>	uomosul.edu.iq					
Name: As	sist. Lec. I	Vais Salim Saadallah					
Email: dr	mais@uom	iosul.edu.iq					
8. Co	ourse Objec	ctives					
Course U	Djectives	for more discourse		anitanina madiaati			
Enable in	e students	for preparing, dispension	ig reviewing and in	onitoring medicatio	on to ensure		
q T_e	aching and	Learning Strategies	IOII				
Strategy	acting and	Lecturing					
Strategy		Seminars					
		Homework					
		Ouiz					
		Practical laboratory demonstrations, experimental calculations and Lab book					
		catalogue					
10. Cou	rse Structu	re					
Week	Hours	Required Learning	Unit or subject	Learning	Evaluation		
		Outcomes	name	method	method		
1	2+2	Relationship between	Dilution and	Theoretical			
		strength and total	concentration	lectures.			
		quantity, dilution and			Paper-based		
		concentration of		Laboratory	exams		
		liquids, dilution and		experiments	Chuing		
		concentration of					
	0.0	solids.					
2	2+2	Some calculations	Dilution	Theoretical			
		about stock solutions	concentration;	lectures.	Paper-based		
			SLOCK SOLUTION	Laboratory	exams		
				demonstration			
1	1	1	1	ucinonstration.			

3	2+2	How to prepare different solutions by using allegation method	Dilution concentration; allegation	Theoretical lectures. Laboratory demonstration.	Paper-based exams
4	2+2	Tutorial and practical problems	Tutorial and pract problems	Theoretical lectures. Laboratory demonstration.	Paper-based exams
5	2+2	Density vs. specific gravity, specific gravity of liquids, specific gravity of solids	Density, specific gravity s specific volume	Theoretical lectures. Laboratory experiments.	Paper-based exams
6	2+2	Problems about density and specific gravity in calculation of weight and volum.	Use of specific gravity in calculati of weight and volume	Theoretical lectures. Laboratory demonstration.	Paper-based exams
7	2+2	Preparation of isotonic solutions with determination of milliequivalents, millimoles and milliosmoles	Isotonic solutions how to prepare isotonic solution at measurement of osmolarity and millimols	Theoretical lectures. Laboratory demonstration.	Paper-based exams
8			Mid-term exan	1	
9	2+2	Tutorial and practical problems	Tutorial and practi problems	Theoretical lectures. Laboratory demonstration.	Paper-based exams
10	2+2	Some calculation for administration of intravenous admixtures and parenteral solutions	Intravenous admixtures and parenteral nutritior	Theoretical lectures. Laboratory demonstration.	Paper-based exams
11	2+2	How to measurement the rate of flow for different intravenous fluids	Rate of flow of intravenous fluids	Theoretical lectures. Laboratory demonstration.	Paper-based exams
12	2+2	Practical problems	Some calculation involving "units," Mg/mg," and other measures of potency	Theoretical lectures. Laboratory demonstration.	Paper-based exams
13	2+2	Dosage calculation based on Creatinine clearance	Some calculation associated with drug availability a pharmacokinetics	Theoretical lectures.	Paper-based exams

		Laboratory demonstration.				
14	Students' seminars					
11. Co	ourse Evaluation					
	 20 M Theoretical assessments; (paper-based mid-term exam + att 20 M practical assessment (attend 60 M paper-based theoretical fina 100 M total 	endance + seminar) lance + quiz + practice+ oral-based exam) l exam				
12. Le	earning and Teaching Resources					
Require	ed textbooks	 Ansel HC, Stoklosa MJ. Pharmaceutical calculations 13th edition Philadelphia, PA: Lippincott. Williams and Wilkins, 2010 Laboratory Manual for Practical Pharmacology adopted by the department. 				
Main re	eferences (sources)	 Ansel HC, Stoklosa MJ. Pharmaceutical calculations ,10th , 13th edition Philadelphia, PA: Lippincott. Williams and Wilkins, 2010 Code of Ethics for Pharmacist. American Pharmaceutical Association 				
Electro	nic References, Websites	https://www.pharmacist.com/				