



## **Lecture title: Lipids**

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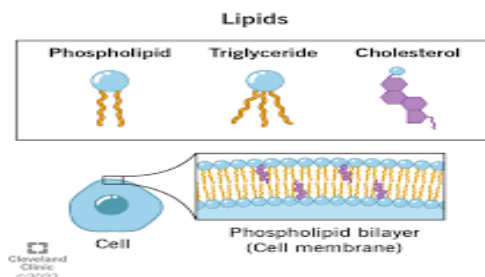
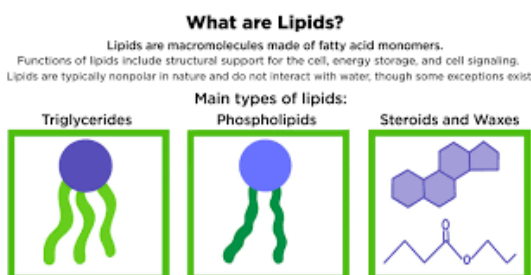
**Summary:** Lipids are a major source of energy for the body besides their various other biochemical function and their role in cellular structure. Lipids are a heterogenous group of water insoluble (hydrophobic) organic molecules. Lipids include fats, oils, steroids, waxes and related compounds. Lipids are water insoluble components of cells.

- Some lipids serve as structural component of membrane and others as storage form of fuel.
- Triacylglycerols contain three fatty acid molecules esterified to the three hydroxyl groups of the glycerol and are storage fats.
- The polar lipids (phospholipids, glycolipids and cholesterol), which have polar heads and nonpolar tails, are amphipathic and major components of membranes..



## What are Lipids?

Lipids are a major source of energy for the body, besides their various other biochemical functions and their role in cellular structure. Lipids are a heterogeneous group of water-insoluble (hydrophobic) organic molecules. Lipids include fats, oils, steroids, waxes, and related compounds.



## Definition of Lipids

Lipids may be defined as organic substances insoluble in water but soluble in organic solvents like chloroform, ether, and benzene.



## Classification of Lipids

The most commonly classification of lipids is:

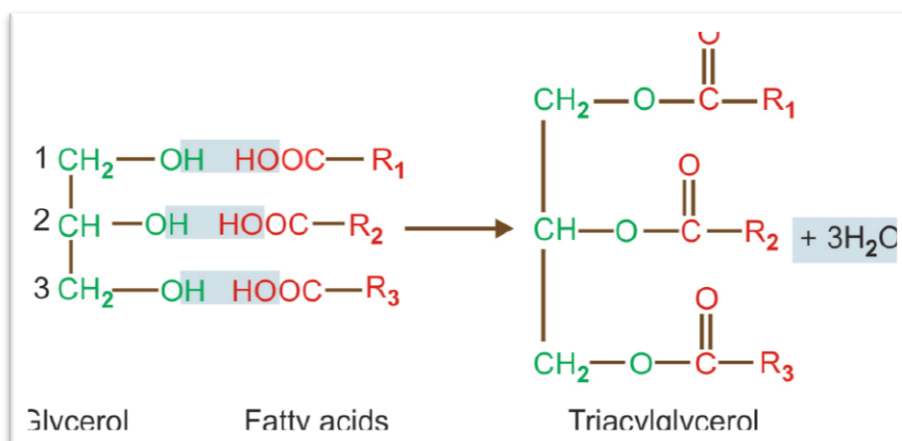
1. Simple lipids
2. Complex or compound lipids
3. Derived lipids

**Simple Lipids:** These are esters of fatty acids with various alcohols.

Depending on the type of alcohols, these are sub classified as:

1. Neutral fats or triacylglycerol or triglycerides.
2. Waxes.

**Neutral fats or triacylglycerol or triglycerides:** These are esters of fatty acids with alcohol **glycerol** (fig.1) . Because they are **uncharged**, they are termed as **neutral fat**.



**Fig.1: Glycerol and triacylglycerol**



A fat in **liquid state** is called an **oil**, e.g. vegetable oils like groundnut oil, mustard oil, corn oil, etc.

**Waxes:** These are esters of fatty acids with **higher molecular** weight monohydric long chain alcohols. For example: Bees-wax

These are widely used in pharmaceutical, cosmetic and other industries in the manufacture of lotions, ointments and polishes.