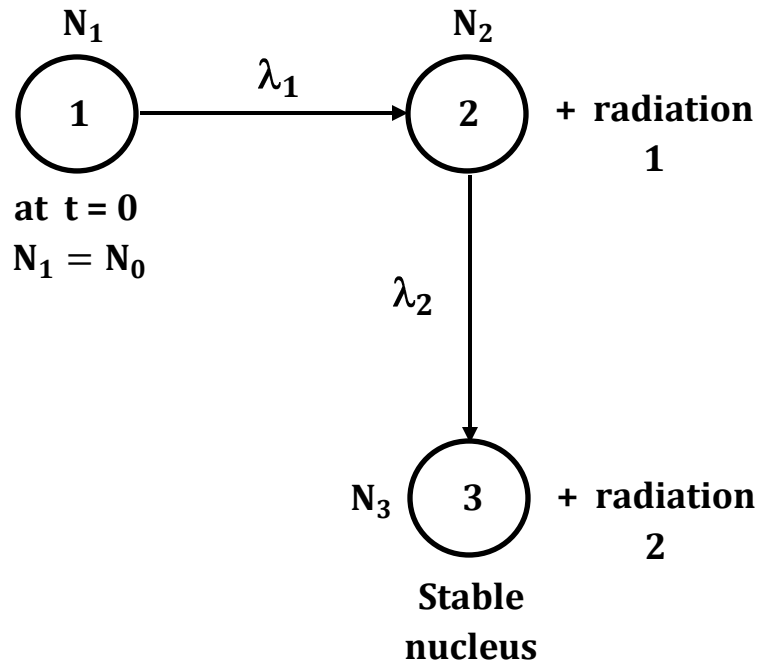


Production of Radioisotope by a Decaying Parent

(Radioactive – Series Decay)

Suppose a parent nucleus (1) decays and produced radioactive daughter (2) and radiation (1), and that the daughter (2) producing a stable nucleus (3) and radiation (2). Note that every decaying nucleus produces one daughter nucleus.



$$\frac{dN_1}{dt} = -\lambda_1 N_1 \quad \text{-----} \quad (1)$$

$$\frac{dN_2}{dt} = \lambda_1 N_1 - \lambda_2 N_2 \quad \text{-----} \quad (2)$$

$$\frac{dN_3}{dt} = \lambda_2 N_2 \quad \text{-----} \quad (3)$$

at time $t = 0$

$$N_1 = N_0$$

$$N_2 = N_3 = 0$$

at time $t = \infty$

$$N_1 = N_2 = 0$$

$$N_3 = N_0$$

if N_0 is the original number of parent nucleus (1) at time ($t = 0$) then :

$$N_1 = N_0 e^{-\lambda_1 t} \quad \text{-----} \quad (4)$$

From eq. (2) and (4) we have :

$$\frac{dN_2}{dt} = +\lambda_2 N_2 = N_0 e^{-\lambda_1 t} \quad \text{-----} \quad (5)$$

if daughter (2) is stable nucleus (i.e. $\lambda_2 = 0$) then $N_2 = N_0(1 - e^{-\lambda_1 t})$ at any time (t) :

$$N_1 + N_2 = N_0$$

The total number of nuclei constant

This is a differential equation consist of a general solution homogeneous solution equation. The solution for N_2 as a function of time will be :

$$N_2 = N_0 (C_1 e^{-\lambda_1 t} + C_2 e^{-\lambda_2 t}) \quad \text{-----} \quad (6)$$

In order to evaluate the coefficients C_1 and C_2 we substitute N_2 and $\frac{dN_2}{dt}$ from eq. (6) in eq. (5).

The differentiation of eq. (6) give :

$$\frac{dN_2}{dt} = N_0 (-\lambda_1 C_1 e^{-\lambda_1 t} - \lambda_2 C_2 e^{-\lambda_2 t})$$

Substituting into eq. (5), we obtain :

$$\cancel{N_0} (-\lambda_1 C_1 e^{-\lambda_1 t} - \lambda_2 C_2 e^{-\lambda_2 t}) + \lambda_2 \cancel{N_0} (C_1 e^{-\lambda_1 t} + C_2 e^{-\lambda_2 t}) = \lambda_1 \cancel{N_0} e^{-\lambda_1 t}$$

$$\therefore -\lambda_1 C_1 e^{-\lambda_1 t} - \lambda_2 C_2 e^{-\lambda_2 t} + \lambda_2 C_1 e^{-\lambda_1 t} + \lambda_2 C_2 e^{-\lambda_2 t} = \lambda_1 e^{-\lambda_1 t}$$

$$e^{-\lambda_1 t} (-\lambda_1 C_1 + \lambda_2 C_1 - \lambda_1) = 0 \quad \text{-----} \quad (7)$$

Since $e^{-\lambda_1 t} \neq 0$, then $(-\lambda_1 C_1 + \lambda_2 C_1 - \lambda_1) = 0$

$$\therefore C_1 = \frac{\lambda_1}{\lambda_2 - \lambda_1}$$

The coefficient C_2 depends on the value N_2 , at time $t = 0$, $N_2 = 0$. From eq. (6)

$0 = N_0(C_1 + C_2)$, and since $N_0 \neq 0$, then

$$C_1 + C_2 = 0 \quad \text{or} \quad C_2 = -C_1 = -\frac{\lambda_1}{\lambda_2 - \lambda_1}$$

then the total number of daughter nuclei (N_2) will be :

$$N_2 = N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} (e^{-\lambda_1 t} - e^{-\lambda_2 t}) \quad \text{-----} \quad (8) \quad \text{معادلة اساس}$$

The activity of radioactive daughter (2) is $A_2 = \lambda_2 N_2$

$$\therefore A_2 = \lambda_2 N_2 = \lambda_2 N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} (e^{-\lambda_1 t} - e^{-\lambda_2 t}) \quad \text{-----} \quad (9)$$

$$A_2 = (N_0 \lambda_1) \frac{\lambda_2}{\lambda_2 - \lambda_1} (e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

And since the activity of the parent (1) at time (t) is

$$A_1 = (N_1 \lambda_1) = (N_0 \lambda_1) e^{-\lambda_1 t}$$

$$\therefore (N_0 \lambda_1) = (N_1 \lambda_1) e^{+\lambda_1 t}$$

\therefore The daughter activity

$$A_2 = A_1 \frac{\lambda_2}{\lambda_2 - \lambda_1} [1 - e^{-(\lambda_2 - \lambda_1)t}] \text{ ----- } (10)$$

The total number of nucleus 3 (i.e. N_3) can be obtained from eq. (8) and (3) if $N_3 = 0$ at time $t = 0$:

$$N_3 = N_0 \frac{\lambda_1 \lambda_2}{\lambda_2 - \lambda_1} \left(\frac{1 - e^{-\lambda_1 t}}{\lambda_1} - \frac{1 - e^{-\lambda_2 t}}{\lambda_2} \right)$$

* if parent (1) is **short - lived** compared to the daughter (2), i.e. $T_1 \ll T_2$ or $(\lambda_1 \gg \lambda_2)$, then after a **long - time** ($t \gg \frac{1}{\lambda_1}$) ; $e^{-\lambda_1 t} \ll e^{-\lambda_2 t}$, and from equation (8) we have :

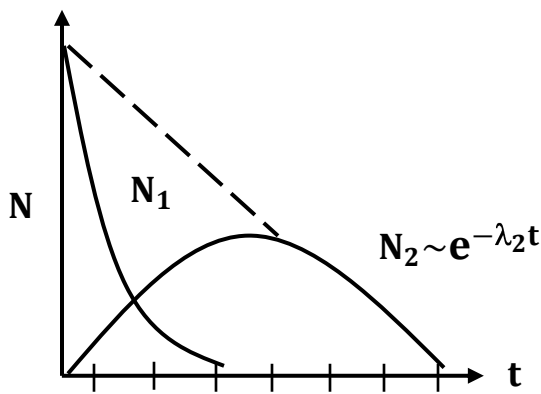
$$N_2 = N_0 \frac{\lambda_1}{\lambda_1 - \lambda_2} (e^{-\lambda_2 t})$$

This means that the decay of daughter (2) is determined by its own **half - life** T_2 .

* if the parent (1) is **long - lived** compared to its daughter (2) i.e. $T_1 \gg T_2$ or $(\lambda_1 \ll \lambda_2)$, then after a **long - time** ($t \gg \frac{1}{\lambda_1}$) , $e^{-\lambda_2 t} \approx 0$ and N_2 from eq. (8) will be :

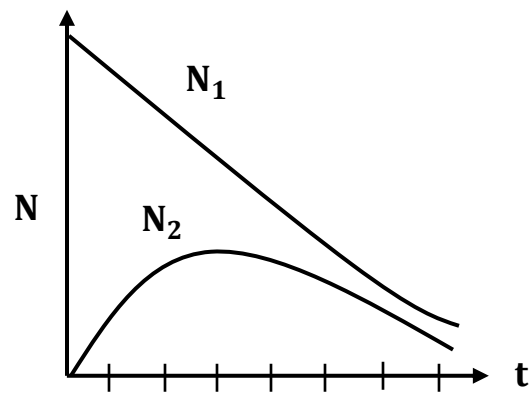
$$N_2 = N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} (e^{-\lambda_1 t})$$

This means that the decay of daughter (2) is determined by the **half - life** of parent (1) .



Short - lived Parent

$$T_1 < T_2 ; (\lambda_1 > \lambda_2)$$



Long - lived Parent

$$T_1 > T_2 ; (\lambda_1 < \lambda_2)$$

Fig. Decay of Radioactive Daughter

Transient Equilibrium : if the parent is **longer – lived** than the daughter ($T_1 > T_2$) but the **half – life** of parent is not very long , then $e^{-\lambda_2 t} \ll e^{-\lambda_1 t}$, therefore ;

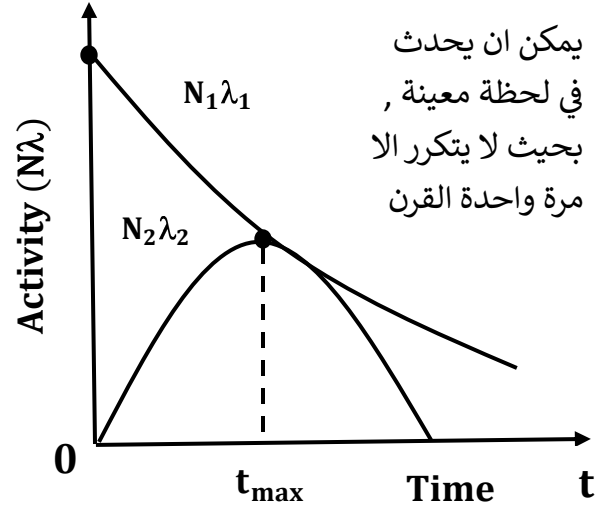
$$N_2 = N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} e^{-\lambda_1 t} = (N_0 e^{-\lambda_1 t}) \frac{\lambda_1}{\lambda_2 - \lambda_1} = N_1 \frac{\lambda_1}{\lambda_2 - \lambda_1}$$

$$\frac{N_2}{N_1} = \frac{\lambda_1}{\lambda_2 - \lambda_1} \quad . \quad \text{Multiplying by } \left(\frac{\lambda_2}{\lambda_1}\right) \text{ we obtain } \frac{N_2 \lambda_2}{N_1 \lambda_1} = \frac{\lambda_2}{\lambda_2 - \lambda_1}$$

$$\therefore \frac{\text{Activity of 2}}{\text{Activity of 1}} = \frac{\lambda_2}{\lambda_2 - \lambda_1} \quad \text{or} \quad \frac{A_2}{A_1} = \frac{\lambda_2}{\lambda_2 - \lambda_1}$$

Secular (Ideal) Equilibrium :

When $T_1 \gg T_2$, i.e. when $\lambda_1 \ll \lambda_2$, the activity of the residual parent ($\lambda_1 N_1$) and of the accumulated daughter ($\lambda_2 N_2$) are equal since from transient equilibrium we have :



$$\frac{\text{Activity of 2}}{\text{Activity of 1}} = \frac{A_2}{A_1} = \frac{\lambda_2 N_2}{\lambda_1 N_1} = \frac{\lambda_2}{\lambda_2 - \lambda_1}$$

and since $\lambda_1 \ll \lambda_2$, therefore $\lambda_2 - \lambda_1 \approx \lambda_2$

$$\therefore \frac{\text{Activity of 2}}{\text{Activity of 1}} = 1 ; \text{ i.e. } \text{Activity of daughter 2} = \text{Activity of Parent 1}$$

Note that eq. (2) shows immediately that $dN_2/dt = 0$

$$\frac{dN_2}{dt} = \lambda_1 N_1 - \lambda_2 N_2 = 0 \quad \therefore \lambda_2 N_2 = \lambda_1 N_1$$

In this case the equilibrium is called “**Secular**” or “**Ideal**” (in which type 2 nuclei is decaying at the same rate in which is formed) and exist only at time when the activity of daughter (2) is maximum, which is t_{\max} .

Time of Maximum Activity of Daughter Product

From eq. (8) the amount of N_2 is zero both at $t = 0$ and $t = \infty$, when all the atoms of both parent (1) and daughter (2) have decayed. At some intermediate time t_{\max} , the amount of daughter (2) and hence its activity ($\lambda_2 N_2$), pass through a maximum value. $dN_2/dt = 0$ for time (t_{\max}). From differential of eq. (8) we find (t_{\max}) for maximum activity of (N_2).

$$N_2 = N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} (e^{-\lambda_1 t} - e^{-\lambda_2 t})$$

$$\therefore \frac{dN_2}{dt} = N_0 \frac{\lambda_1}{\lambda_2 - \lambda_1} (-\lambda_1 e^{-\lambda_1 t} + \lambda_2 e^{-\lambda_2 t}) = 0$$

$$\therefore -\lambda_1 e^{-\lambda_1 t_{\max}} + \lambda_2 e^{-\lambda_2 t_{\max}} = 0 \quad ; \quad \frac{\lambda_2}{\lambda_1} = e^{(\lambda_2 - \lambda_1) t_{\max}}$$

$$\text{Therefore ; } t_{\max} = \frac{\ln (\lambda_2 / \lambda_1)}{\lambda_2 - \lambda_1} \quad \text{or} \quad t_{\max} = \tau_2 \left(\frac{T_1}{T_1 - T_2} \right) \ln \left(\frac{T_1}{T_2} \right)$$

This important result shows that t_{\max} is positive and real for either $T_1 > T_2$ or $T_1 < T_2$.

Production (Yield) of Radioactive Nuclei by Nuclear Bombardment

Let's assume that we place a target of stable nuclei into a reactor or an accelerator such as a cyclotron. The nuclei of the target will capture (**absorb**) a neutron or a charged particle to produce a radioactive species.

Suppose that **Q** atoms (**Nuclei**) are produced per second at constant rate

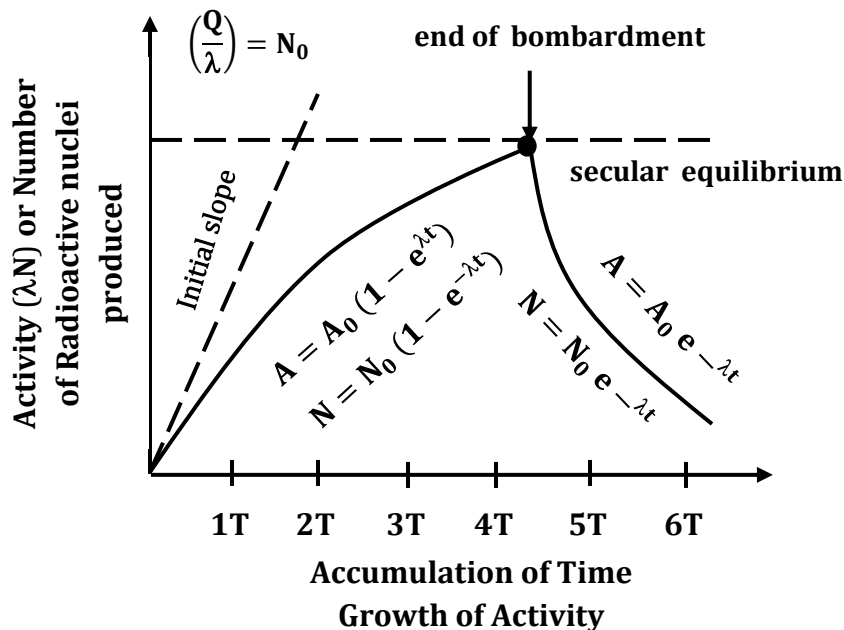
$$Q = N_0 \sigma \Phi = \text{constant}$$

Where

$N_0 \Rightarrow$ target nuclei

$\sigma \Rightarrow$ cross - section , i.e. probability production of radioactive nuclei (capture probability).

$\Phi \Rightarrow$ flux of the incident beam (neutron / cm².sec).



And let **N** be the number of radioactive atoms present at time **t** . The net rate of increase of radioactive nuclei

Net Rate = Rate of production – Rate of decay

$$\frac{dN}{dt} = Q - \lambda N$$

$$\frac{dN}{dt} + \lambda N = Q$$

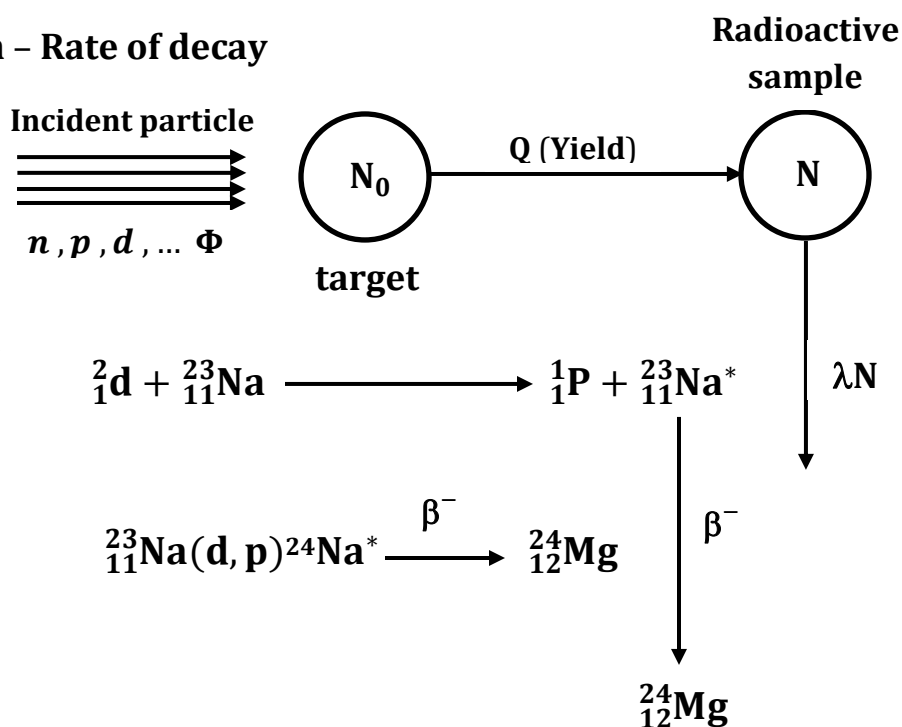
Multiplying by $e^{+\lambda t}$, we obtain

$$e^{+\lambda t} \frac{dN}{dt} + \lambda N e^{+\lambda t} = Q e^{+\lambda t}$$

$$\frac{d}{dt} (N e^{\lambda t}) = Q e^{+\lambda t}$$

By integration , we have

$$N e^{\lambda t} = \frac{Q}{\lambda} e^{\lambda t} + C \text{ or}$$



$N = \frac{Q}{\lambda} + C e^{-\lambda t}$. the integration constant C can be determined from the condition that : when $t = 0$, $N = 0$, which gives : $C = -\frac{Q}{\lambda}$

$$\therefore N = \frac{Q}{\lambda} - \frac{Q}{\lambda} e^{-\lambda t} \quad \text{or} \quad N = \frac{Q}{\lambda} (1 - e^{-\lambda t}) \quad ; \quad N_0 = \frac{Q}{\lambda} = Q\tau$$

$$\therefore N = N_0 (1 - e^{-\lambda t})$$

N approaches approximately a limiting value of $\frac{Q}{\lambda}$, it reaches **50%** of this value after **one half - life (1T)** , **75%** after **(2T)** and \approx **88%** after **(3T)** **three half - lives**. Since $N = \frac{Q}{\lambda} (1 - e^{-\lambda t})$, then $N\lambda = Q(1 - e^{-\lambda t})$ or the activity A will be : $A = Q(1 - e^{-\lambda t})$, which is the strength of the radioactive source immediately after the end of its production from the curve of growth of activity , it is not **practical** to bombard the target for longer than **2 to 3 half - lives** since **75%** or **88%** of the maximum number or (of the maximum activity) of radioactive nuclei produced.

Note that for irradiation times long compared with the **half - life** ($t \gg T_{1/2}$) , the exponential ($e^{-\lambda t}$) approaches **Zero** and the activity is approximately constant : ($A(t) = Q$ at $t \gg T_{1/2}$).

In this case new activity is being formed at the same rate at which the older activity decays. This is an example of secular equilibrium.