

(2-3) Pressure

Pressure is defined as the magnitude of the normal force acting per unit surface area. The pressure is thus a scalar quantity. We write this mathematically as

$$P = \frac{F}{A} \quad \dots \dots \dots (2 - 2)$$

The SI unit for pressure is **newton/meter² (N/m²)**, which is given the special name **Pascal (Pa)**, in honor of the French mathematician, physicist, and philosopher, Blaise Pascal. Hence, **1 Pa = 1 N/m²**.

There are others used units like **lb/in² (psi)**, **dyne/cm²**, **bar**, **atm**, and **mm.Hg (torr)** the relation between them is:

$$1 \text{ N/m}^2 = 1 \text{ Pa}$$

$$1 \text{ N/m}^2 = 1.45 \times 10^{-4} \text{ lb/in}^2$$

$$1 \text{ N/m}^2 = 9.87 \times 10^{-6} \text{ atm}$$

$$1 \text{ N/m}^2 = 10^{-5} \text{ bar}$$

$$1 \text{ N/m}^2 = 10 \text{ dyne/cm}^2$$

$$1 \text{ atm} = 760 \text{ mm.Hg} = 1.013 \times 10^5 \text{ Pa}$$

Pressures are not limited to fluids, as the following examples show.

Example 5: A man weighs **200 lb**. At one particular moment when he walks, his right heel is the only part of his body that touches the ground. If the heel of his shoe measures **9 cm** by **8 cm**, what pressure does the man exert on the ground in **Pa**?

Answer:

$$1 \text{ cm} = 0.393 \text{ in}$$

$$\text{So, } 9 \text{ cm} = 3.5 \text{ in}$$

$$8 \text{ cm} = 3.25 \text{ in}$$

$$P = \frac{F}{A}$$

$$= \frac{200}{3.5 \times 3.25} = 17.6 \text{ lb/in}^2$$

$$= 1.21 \times 10^5 \text{ Pa}$$

Example 6: A **100 lb** woman is wearing “high-heel” shoes. The cross section of her high-heel shoe measures **1.27 cm** by **1.59 cm**. At a particular moment when she is walking, only one heel of her shoe makes contact with the ground. What is the pressure exerted on the ground by the woman in **Pa**?

Example 7: The two feet of a **60 kg** person cover an area of **500 cm²**, (a) Determine the pressure exerted by the two feet on the ground. (b) If the person stands on one foot, what will the pressure be under that foot?

Example 8: A man has a **140 kg** regular bed that is supported by four legs each leg has a circular cross section of radius **2 cm**. What pressure does this bed exert on the floor?

Example 9: A child weighs **364 N** and sits on a three-legged stool, which weighs **41 N**. The bottoms of the stool's legs touch the ground over a total area of **19.3 cm²**.

- a) What is the average pressure that the child and stool exert on the ground?
- b) How does the pressure change when the child leans over so that only two legs of the stool touch the floor?

(2-4) Pressure in Fluid of uniform density

Pressure exerted by a fluid is easily determined with the aid of figure 2-1, which represents a pool of water. We want to determine the pressure p at the bottom of the pool caused by the water in the pool. By our definition, equation (2-2), the pressure at the bottom of the pool is the magnitude of the force acting on a unit area of the bottom of the pool. But the force acting on the bottom of the pool is caused by the weight of all the water above it. Thus,

$$P = \frac{F}{A} = \frac{\text{weight of water}}{\text{Area}} = \frac{w}{A} = \frac{mg}{A} \quad \dots \dots \dots (2 - 3)$$

The mass of the water in the pool, given by equation (2-1), is ($m = \rho v$). The volume of all the water in the pool is just equal to the area A of the bottom of the pool times the depth h of the water in the pool, that is,

$$v = Ah \quad \dots \dots \dots (2 - 4)$$

Substituting equations (2-1) and (2-4) into equation (2-3) gives for the pressure at the bottom of the pool:

$$P = \frac{mg}{A} = \frac{\rho v g}{A} = \frac{\rho A h g}{A}$$

Thus,

$$P = \rho g h \quad \dots \dots \dots (2 - 5)$$

Equation (2-5) says that the water pressure at any depth h in any pool is given by the product of the density of the water in the pool, the acceleration due to gravity g , and the depth h in the pool. Equation (2-5) is sometimes called the **hydrostatic equation**.

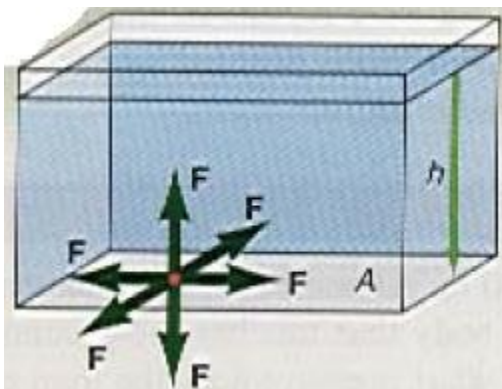


Figure 2-1 water pool

(2-5) Measurement of Pressure

(2-5-1) Atmospheric Pressure

The pressure of the air in the atmosphere was first measured by Evangelista Torricelli (1608-1647), a student of Galileo, by the use of a mercury **barometer**. A long narrow tube is filled to the top with mercury, chemical symbol Hg. It is then placed upside down into a reservoir filled with mercury, as shown in figure (2-2).

The mercury in the tube starts to flow out into the reservoir, but it comes to a stop when the top of the mercury column is at a height h above the top of the mercury reservoir, as also shown in figure (2-2). The mercury does not empty completely because the normal pressure of the atmosphere p_o pushes downward on the mercury reservoir. Because the force caused by the pressure of a fluid is the same in all directions, there is also force acting upward inside the tube at the height of the mercury reservoir, and hence there is also a pressure p_o acting upward as shown in figure (2-2). This force upward is capable of holding the weight of the mercury in the tube up to a height h . Thus, the pressure exerted by the mercury in the tube is exactly balanced by the normal atmospheric pressure on the reservoir, that is,

$$P_o = P_{Hg} \quad \dots \dots \dots (2-6)$$

But the pressure of the mercury in the tube P_{Hg} , given by equation (2-5),

$$P_{Hg} = \rho_{Hg}gh \quad \dots \dots \dots (2-7)$$

Substituting equation (2-7) back into equation (2-6), gives

$$P_o = \rho_{Hg}gh \quad \dots \dots \dots (2-8)$$

Equation (2-8) says that normal atmospheric pressure can be determined by measuring the height h of the column of mercury in the tube. It is found experimentally, that on the average, normal atmospheric pressure can support a column of mercury 76.0 cm high, or 760 mm high. The unit of 1.00 mm of Hg is sometimes called a torr in honor of Torricelli. Hence, normal atmospheric pressure can also be given as 760 torr. Using the value of the density of mercury of $1.360 \times 10^4 \text{ kg/m}^3$, normal atmospheric pressure, determined from equation (2-8), is

$$\begin{aligned} P_o &= \rho_{Hg}gh = \left(1.36 \times 10^4 \frac{\text{kg}}{\text{m}^3}\right) \left(9.8 \frac{\text{m}}{\text{s}^2}\right) (0.760 \text{ m}) \\ &= 1.013 \times 10^5 \text{ N/m}^2 (\text{Pa}) \end{aligned}$$

Thus, the average or normal atmospheric pressure acting on us at the surface of the earth is $1.013 \times 10^5 \text{ Pa}$.

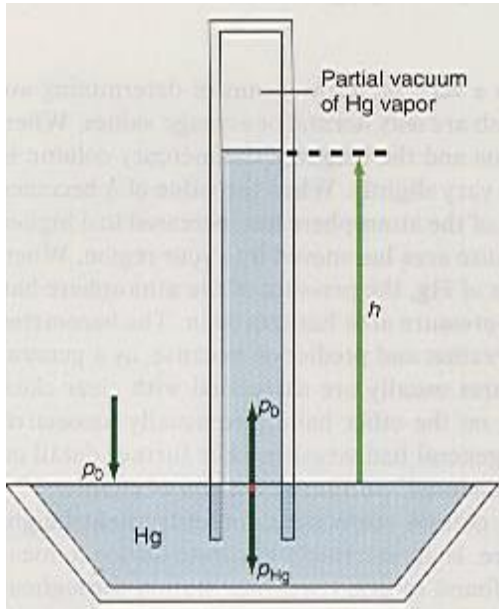


Figure 2-2 A mercury barometer

